

# Mössbauer study of the Heusler-type $\text{Fe}_2\text{MAI}$ compounds for $\text{M} = \text{V}, \text{Cr}, \text{Fe}, \text{Co}, \text{Ni}$

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**Abstract** The crystal structure and magnetic properties of  $\text{Fe}_2\text{MAI}$  compounds for  $\text{M} = \text{V}, \text{Cr}, \text{Fe}, \text{Co}$  and  $\text{Ni}$  were investigated by X-ray, magnetostatic and Mössbauer effect methods. It was found that the investigated compounds possess not perfectly ordered fcc structure of the  $\text{L}_{2_1}$ -type. With increasing number of  $3d$  electrons of  $\text{M}$  component the transition from the paramagnetic to ferromagnetic behaviour follows. Mössbauer spectra were analysed by a hyperfine field distribution method. They are characteristic of the ferromagnetic phase for  $\text{M} = \text{Fe}, \text{Co}$  and  $\text{Ni}$  and indicated the early observed preference of these atoms for the Fe sites.

**Key words** Heusler alloys • Mössbauer spectroscopy • site preference • magnetostatic measurements

## Introduction

Heusler-type alloys are cubic ternary compounds with the formula  $\text{X}_2\text{YZ}$ , where  $\text{X}$  and  $\text{Y}$  are two different transition metals and  $\text{Z}$  is one  $sp$  element.

They can crystallize in ordered fcc cubic structure of the  $\text{L}_{2_1}$ -type. Three elements  $\text{X}, \text{Y}, \text{Z}$  occupy crystallographically non-equivalent positions [12]. Figure 1 shows the unit cell of such structure.

These alloys have been widely investigated because they exhibit very interesting magnetic and electric properties [1, 2, 7–11, 13, 16, 17]. Iron based transition metal alloys show a great sensitivity to effect of local environment [1, 6, 7, 11, 14, 15]. The tendency for partial disordering of the crystal structure is often observed [7, 8, 11, 14, 15].

In this work we present results of X-ray (XRD), magnetostatic and Mössbauer investigations of the Heusler-type compounds  $\text{Fe}_2\text{MAI}$  with  $\text{M} = \text{V}, \text{Cr}, \text{Fe}, \text{Co}$  and  $\text{Ni}$ .

## Experimental

Polycrystalline  $\text{Fe}_2\text{MAI}$  compounds with  $\text{M} = \text{V}, \text{Cr}, \text{Fe}, \text{Co}$  and  $\text{Ni}$  were obtained by the Czochralski method from a levitated melt using high purity starting materials. After melting, the ingots were powdered to a particle size less than  $30 \mu\text{m}$  and vacuum annealed to obtain the  $\text{L}_{2_1}$ -type structure. Annealing procedure was similar to that presented in Ref. [2]. XRD measurements of all samples was performed at room temperature using a Siemens D5000 X-ray diffractometer and  $\text{Cu K}_\alpha$  radiation.

The magnetic measurements of the investigated samples were performed at a temperature up to  $800 \text{ K}$  in

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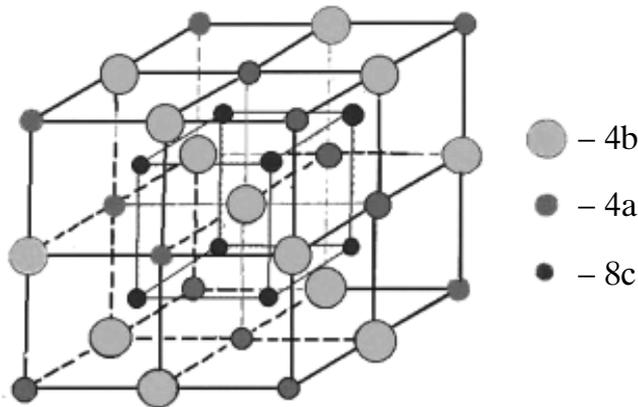


Fig. 1. The unit cell of the  $L2_1$ -type structure [12].

the magnetic field up to 12 kOe (1.2 T) using the Faraday method. The Curie temperature ( $T_C$ ) for the investigated compounds was determined from measurements of the magnetization against the temperature in magnetic fields up to 0.1 T and also from Arrott plots. The saturation magnetization at a given temperature was estimated by extrapolation of the magnetization vs. reciprocal magnetic field intensity  $H$  to  $H^{-1} = 0$ .

The Mössbauer measurements were performed at room temperature using a constant acceleration spectrometer with a  $^{57}\text{Co}:\text{Pd}$  source. These Mössbauer spectra were calculated by means of a discrete analysis (a single line and a few Zeeman sextets) and a hyperfine magnetic field distribution using the Hesse-Rübartsch procedure [3].

## Results and discussion

XRD measurements showed that all investigated compounds were single phase and exhibited the fcc structure of the  $L2_1$ -type. Detailed analysis of intensity of the superstructure Bragg reflections showed partial atomic disorder of the crystal structure and a preference of the Co and Ni atoms to occupy the 8c sites. Results of crystallographic and magnetic investigations are presented in Table 1.

The temperature variations of magnetization for selected investigated compounds are presented in Fig. 2.

Magnetic studies showed that the  $\text{Fe}_2\text{VAl}$  compound above 4.2 K does not exhibit a magnetic transition. For the  $\text{Fe}_2\text{CrAl}$  compound, the magnetic transition extends over a wide temperature range. It was found that it disappears with increasing magnetic field intensity. The temperature variation of magnetization for the  $\text{Fe}_2\text{FeAl}$  compound is

Table 1. Lattice constant  $a$ , magnetic moment  $\mu$  at 300 K and the Curie temperature  $T_C$  for the investigated compounds.

Compounds	$a$ [Å]	$\mu$ [ $\mu_B/\text{f.u.}$ ]	$T_C$ [K]
$\text{Fe}_2\text{VAl}$	$5.760 \pm 0.005$	Paramagnetic above 4.2 K	
$\text{Fe}_2\text{CrAl}$	$5.791 \pm 0.005$	$1.67 \pm 0.01$	$330 \pm 5$
$\text{Fe}_2\text{FeAl}$	$5.791 \pm 0.005$	$5.00 \pm 0.01$	$755 \pm 2$
$\text{Fe}_2\text{CoAl}$	$5.766 \pm 0.005$	$4.91 \pm 0.01$	$>800$
$\text{Fe}_2\text{NiAl}$	$5.741 \pm 0.005$	$4.00 \pm 0.01$	$>800$

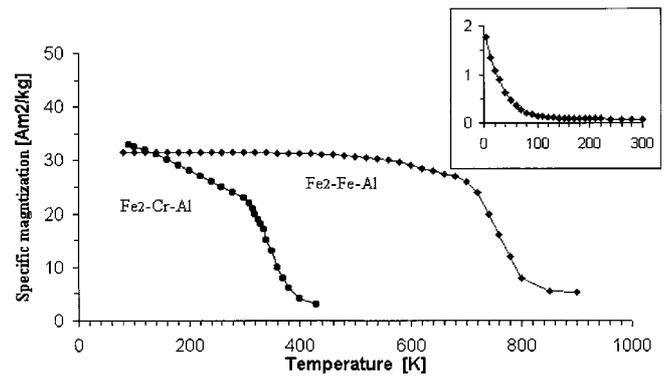


Fig. 2. Magnetization of the  $\text{Fe}_2\text{FeAl}$ ,  $\text{Fe}_2\text{CrAl}$  and  $\text{Fe}_2\text{VAl}$  (insert) compounds vs. temperature measured in  $H = 0.05$  T.

typical for ferromagnets. Figure 3 shows magnetic isotherms at room temperature for the investigated compounds.

The  $\text{Fe}_2\text{CrAl}$  compound achieves the magnetic saturation at much lower magnetic field than the remaining compounds. This allows to suppose that this compound possesses a negligible magnetic anisotropy. The compounds with  $M = \text{Fe}, \text{Co}$  and  $\text{Ni}$  saturated in magnetic fields higher than 5 kOe (0.5 T). A detailed analysis of the magnetic isotherms showed that the approach to magnetic saturation is mainly determined by magnetic inhomogeneities connected with the atomic disorder of the crystal structure [5].

Figure 4 shows the Mössbauer absorption spectra and hyperfine field distributions for the investigated compounds.

The  $\text{Fe}_2\text{VAl}$  compound is a paramagnet at room temperature and its Mössbauer spectrum shows only a single broad absorption line. The Mössbauer spectrum of the  $\text{Fe}_2\text{CrAl}$  compound indicates the coexistence of paramagnetic and weak ordered ferromagnetic phases. Similar behaviour for that compound with the B2-type structure was also observed [7]. The hyperfine field distribution of this compound is broad and it indicates a partial atomic disorder of the crystal structure. A wide temperature range of the magnetic transition, observed for the  $\text{Fe}_2\text{CrAl}$  compound, (Fig. 2) and reaching the magnetic saturation in a weak magnetic field (Fig. 3) allowed to suppose the presence of magnetic clusters of Fe atoms with large magnetic moments ( $10^4 \mu_B$ ). From this reason, the

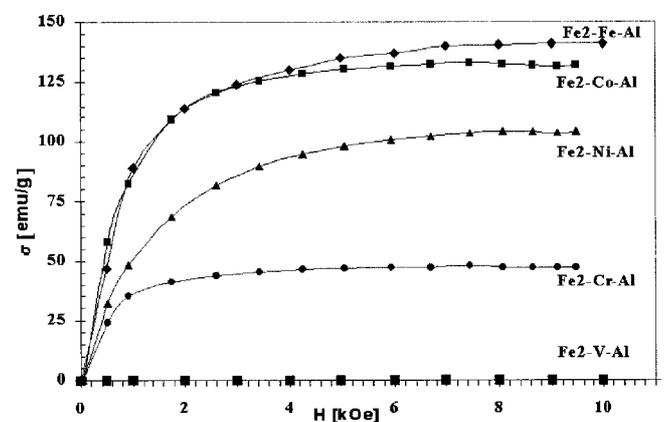


Fig. 3. Magnetic isotherms at room temperature for the investigated compounds.

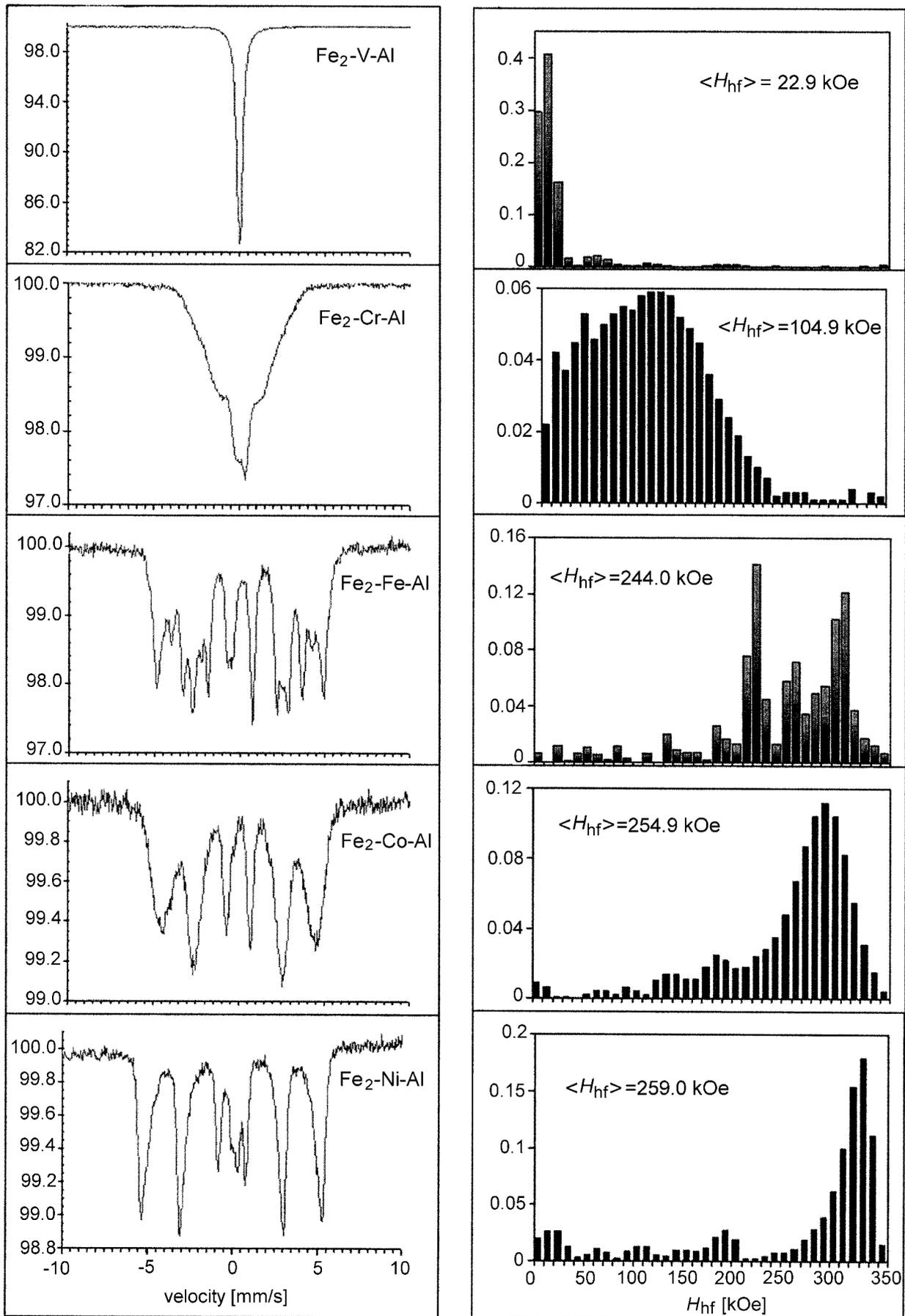
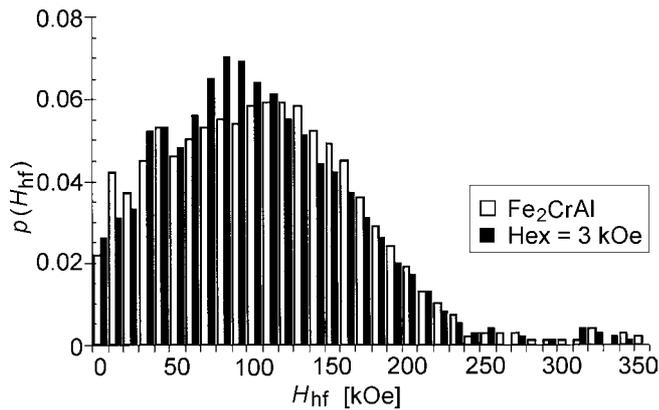


Fig. 4. Mössbauer spectra and hyperfine field distributions for the  $\text{Fe}_2\text{MAl}$  with  $\text{M} = \text{V}, \text{Cr}, \text{Fe}, \text{Co}$  and  $\text{Ni}$ .



**Fig. 5.** The hyperfine field distributions for the  $\text{Fe}_2\text{CrAl}$  compound in the presence (open bars) and absence (full bars) of the external magnetic field.

Mössbauer measurement with an external magnetic field  $H = 3$  kOe perpendicular to gamma rays was done at room temperature. Figure 5 shows a comparison of the hyperfine field distributions in the presence and absence of the external magnetic field for the  $\text{Fe}_2\text{CrAl}$  compound. As can be seen from this picture, the external magnetic field practically does not change the hyperfine field distribution. It means that the magnetic properties of this compound are not connected with Fe magnetic clusters.

The ferromagnetic properties of the compounds  $\text{Fe}_2\text{MAl}$  with  $M = \text{Fe}, \text{Co}$  and  $\text{Ni}$  suggested by magnetic measurements were confirmed by Mössbauer spectra (Fig. 4). Only for the  $\text{Fe}_2\text{NiAl}$  compound, a weak contribution of the paramagnetic phase – a single line – was observed. The hyperfine field distribution of the ferromagnetic compounds points to a partial atomic disorder of the crystal structure. In an ideal ordered  $L2_1$ -type structure, the Fe(I) atom at site 8c has 4 Al at sites 4a and 4 M atoms at sites 4b as near neighbours and it possesses a low magnetic moment. In this situation, only the Zeeman sextet with  $H_{\text{hf}}$  about 220 kOe should be observed. The exchange of an Fe atom with Al or M atoms causes that the Fe(II) atom at site 4a (4b site) is surrounded by 7, 8 Fe(I) atoms and possesses a high magnetic field ( $H_{\text{hf}} = 315\text{--}330$  kOe).

In the  $\text{Fe}_2\text{FeAl}$  compound, the partial atomic disorder causes the hyperfine field distribution for Fe sites 4a and 8c (Fig. 4). As can be seen from Fig. 4 for the  $\text{Fe}_2\text{NiAl}$  and  $\text{Fe}_2\text{CoAl}$  compounds, the amount of Fe atoms at sites 8c decreases, and on the diagram of the hyperfine field distribution the contribution of a high  $H_{\text{hf}}$  component increases. It means that Co and Ni atoms prefer the site 8c. Such tendency was early observed for ternary  $\text{Fe}_{3-x}\text{M}_x\text{Y}$  alloys with  $Y = \text{Al}, \text{Si}$  and  $\text{Ga}$ ,  $M =$  transition metals [4, 6–8, 14, 15].

Investigations of Heusler-like  $\text{Fe}_2\text{MAl}$  compounds with  $L2_1$ -type structure performed in this work show partial

atomic disordering of the crystal structure and confirmed early observed site preference of the M atoms. With increasing number of 3d electrons of M component the transition from the paramagnetic to ferromagnetic behaviour follows.

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